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~~Chapter 7 Periodic Properties of the Elements~~

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The basis of an ionic bond is the__
absorption of water into their solid
structures. Elements tend to gain or lose
electrons in order to acquire the electron
configuration of a__ noble gas. Ions that are
formed from more than one atom are called__

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atoms of each element contained in a single
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CHAPTER 7 REVIEW Chemical Formulas and
Chemical Compounds SECTION 3 SHORT ANSWER
Answer the following questions in the space
provided. 1. Label each of the following
statements as True or False: True a. If the
formula mass of one molecule is x amu, the

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molar mass is x g/mol. False b. Samples of equal numbers of moles of two different chemicals

7 Chemical Formulas and Chemical Compounds
ONLINE Chemistry HMDSscience.com ONLINE LABS include: Test for Fe(II) and Fe(III) Water of Hydration Determining the Empirical Formula of Magnesium Oxide The Formula for Success Chemical Formulas and ... 210 Chapter 7 Untitled-11 210 6/30/2011 12:09:57 PM. aluminum oxide Al O 2 3

CHAPTER 7 Chemical Formulas and Chemical Compounds

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Quiz Modern Chemistry 33 . Name Section Quiz, continued Class Date 6. The Lewis structure of which of the following compounds does not follow the octet rule? a. NH₃ b. BF₃ c. d. H₂O 7. In a double covalent bond, a. one atom has more than eight valence electrons.

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SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

Modern Chemistry 137 Chapter Test Name Class
Date Chapter Test A, continued _____19. In the figure on the previous page, the pH at the equivalence point a. is equal to 7.0. b. is greater than 7.0. c. is less than 7.0. d. cannot be determined from the data given. _____20. In the figure on the previous page, the volume of titration standard

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